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End Semester Examination of Semester-III, 2015

Subject : CHEMISTRY (PG)

Paper : CEM-302 (Org. Spl) (Th)

Full Marks : 40

Time : 2 Hrs

*The figures in the margin indicate the marks
corresponding to the question*

*Candidates are requested to give their answers
in their own word as far as practicable.*

Illustrate the answers wherever necessary.

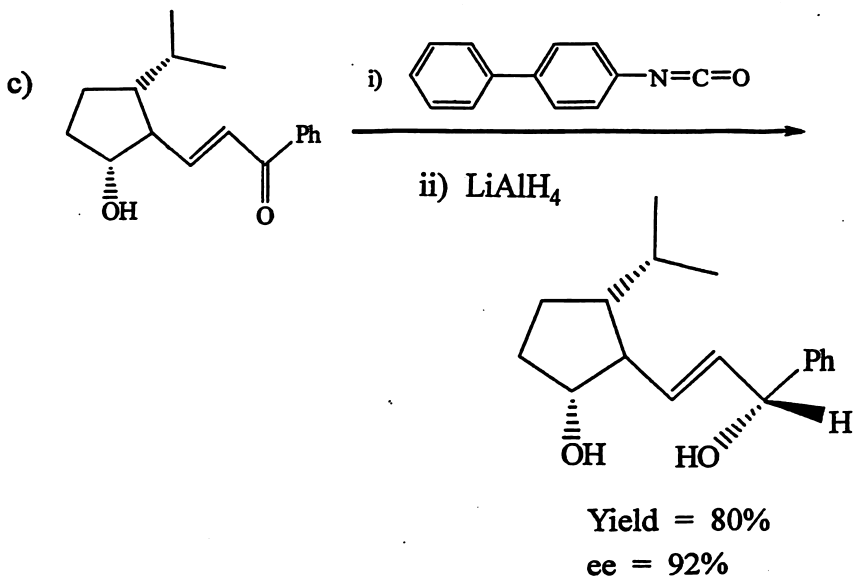
Answer one question from each unit

Unit-I

Answer any one question : 10x1=10

1. a) Show that ester derived from benzoyl formic acid and (R)-2-Octanol produces (R)-atrolactic acid as predominant product when the ester is treated with CH_3MgI followed by hydrolysis.
Would you anticipate any change in the nature of the product when ester of (R)-2-octanol and pyruvic acid is made to react with PhMgBr ? 4
- b) Compare between Cram and Felkin-Anh models regarding dependence of diastereo selectivity on the nature of R-group in nucleophilic addition to RCH(Ph)CH=O . 3

(2)

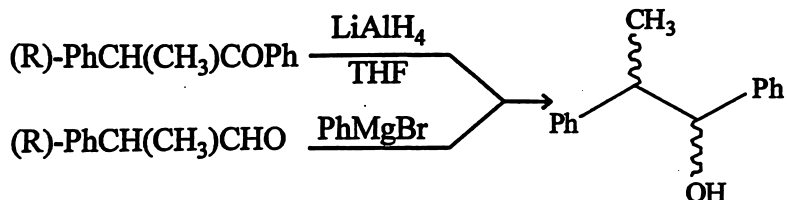


Discuss mechanistically the role of p-diphenyl isocyanate for the stereoselective reduction. 3

2. a) What do you mean by “Stereoselective” and “Stereospecific” reaction? How is the term “asymmetric induction” related to these two reactions? Provide example. 2
- b) $(R)\text{-CH}_3\text{CH}(\text{OMe})\text{CHO}$ reacts with Me_2Zn to give syn- $\text{CH}_3\text{CH}(\text{OH})\text{CH}(\text{OMe})\text{CH}_3$ as major product. Justify or Criticize. 2

(3)

- c) Explain how these reactions give different diastereomeric products. 3



- d) Explain the observation : 3



| Reagents | % Yield (e, e) | % Yield (e, a) | R |
|-------------------------------------|-------------------|-------------------|----------------------------|
| LiAlH_4 | 88 | 12 | H |
| $\text{Li-C}\equiv\text{CH}$ | 88 | 12 | $-\text{C}\equiv\text{CH}$ |
| MeMgBr | 40 | 60 | Me |
| $(\text{Me}_2\text{CH})\text{MgBr}$ | 18 | 82 | CHMe_2 |
| $\text{LiBH}(\text{S-Bu})_3$ | 3.5 | 96.5 | H |

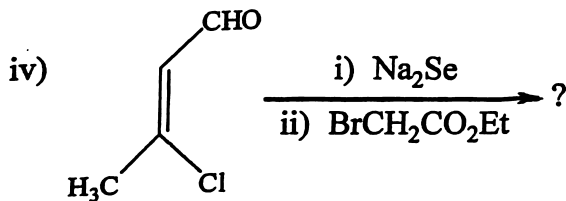
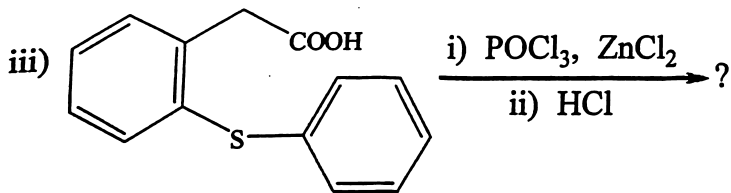
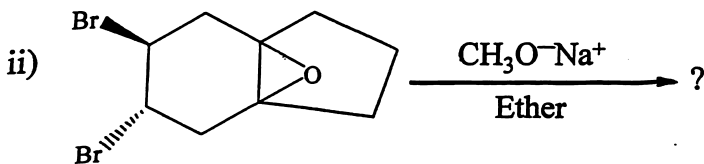
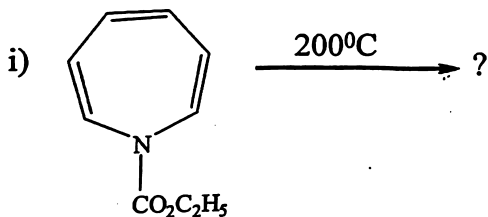
Unit II

Answer any one question : 10X1=10

3. a) Arrange the stability order of 1H, 3H and 4H azepine. Show the hydrogen shift for the transformation 1H-azepine to 3H-azepine. 1+1=2

(4)

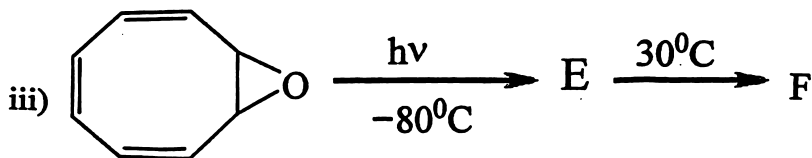
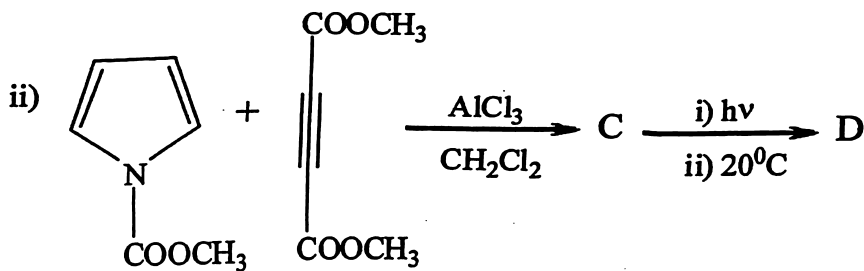
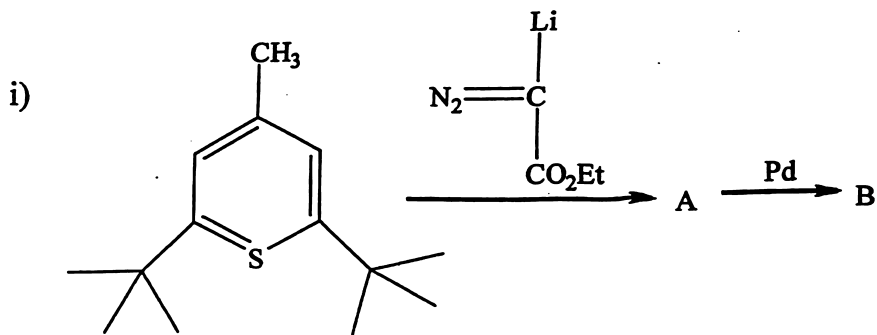
b) Predict the product of the following reactions : 4x2=8



(5)

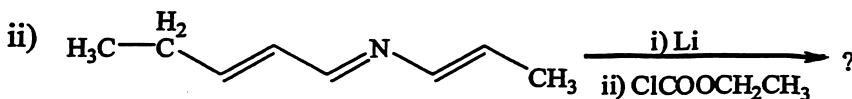
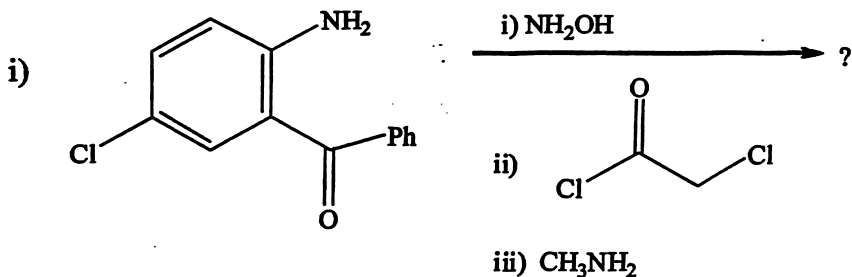
4. a) Identify A, B, C, D, E and F :

6x1=6



(6)

b) Write down the products of the following reactions with mechanism. 3+1=4



Unit III

Answer any one question : 10X1=10

5. a) State the 'biogenetic isoprene rule'. 1

b) Delimitate the following biogenetic conversion :

$$2 \frac{1}{2} \times 2 = 5$$

i) Acetyl coenzyme A to IPP.

ii) IPP to Limonene.

c) How can you prove the presence of α , β unsaturated carbonyl group in Citral. 2

d) Linalool $\xrightarrow{H_3O^+}$ Geraniol

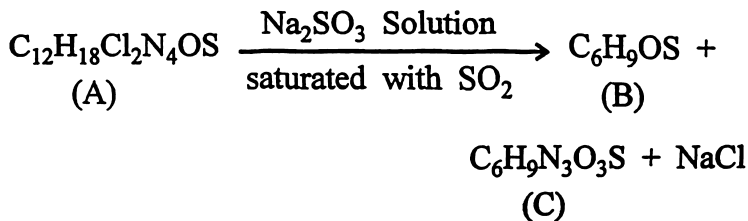
How much information is procured from the above conversion regarding the structure of linalool. 2

(7)

6. a) Outline the steps involved for the biosynthesis of nicotine. 2+2=4
- b) Give two chemical evidences and also reaction sequences in order to prove the presence n-propyl group in coniine. $1\frac{1}{2} \times 2 = 3$
- c) Write mechanism of each step involved in the synthesis of tropinone from succindialdehyde. 3

Unit-IV

7. a) What are vitamins? 1
- b) The following vitamin(A) shows the property as : 3+3+3=9



Chemical and spectral evidence showed that compound (B) is a thiazole and compound (C) is a pyrimidine derivative.

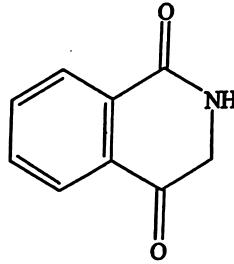
Identify B and C. Establish the structure of A.

8. a) What is antibiotics. Write down chemical structure of norfloxacin and tetracyclines. 1+1=2

(8)

b) Write all the steps for the following synthesis : 4+4=8

i) Penicillin V from



ii) PGF_{2α} from

